INDIAN SCHOOL MUSCAT

CHEMISTRY DEPARTMENT

QUESTION BANK

Surface Chemistry

1.	What do you mean by the term –Adsorption?	1
2.	Explain the terms – Adsorbate and Adsorbent?	1
3.	Give two differences between adsorption and absorption?	1
4.	Why do finely divided solids act as good adsorbents?	1
5.	What is adsorption isotherm?	1
6.	Give an equation showing variation of extent of adsorption with concentration	1
	of a solution?	
7.	What do you mean by the term promoter? Give an example.	1
8.	What is the optimum temperature and pH for enzyme catalysed reactions?	1
9.	What are colloids?	1
10.	Give two examples of solid Sol and Gel?	1
11.	Colloid is a state not a substance. Explain?	1
12.	What are associated colloids? Give an example?	1
13.	What is CMC and Kraft's temperature?	1
14.	Define the term peptization?	1
15.	What is observed when light is passed through a colloidal solution?	1
16.	State Hardy- Schulze rule?	1
17.	What is a protective colloid?	1

18.	Alum is used for purification of water. Why?	1
19.	What are the two types of emulsions?	1
20.	Why does a gas mixed with another gas not form a colloidal system?	1
21.	Explain the terms sorption and desorption.	1
22.	"Chemisorption is highly specific." Illustrate with an example.	1
23.	"Adsorbents in finely divided form are more effective." Why?	1
24.	'Generally high temperature is favourable for chemisorption.' Why?	1
25.	Name the catalyst used in the following process:	1
	(a) Haber's process for the manufacture of NH ₃ gas.	
	(b) Ostwald process for the manufacture of nitric acid.	
26.	Why gas masks are used by miners in coal mines while working?	1
27.	Write the chemical reaction involved in the preparation of sulphur sol.	1
28.	Name the enzyme which converts milk into curd.	1
29.	What are the optimum temperature and pH at which enzymes are highlyactive.	1
30.	What is the composition of colloidion solution?	1
31.	Why do colloidal particles show Brownian movement?	1
32.	Why does sky appear blue to us?	1
33.	What happens when hydrated ferric oxide and arsenioussulphide sols are	1
	mixed in almost equal proportions?	
34.	Gelatin is generally added to ice-cream. Why?	1
35.	Mention the two conditions for the formation of micelles.	1
36.	How is Brownian movement responsible for the stability of sols?	1
37.	Which of the following is more effective in coagulating positively charged	1

	(i) KCl (ii) CaSO ₄ (iii) K ₃ [Fe(CN) ₆].	
38.	Mention one use of ZSM-5 catalyst.	1
39.	Define the term – desorption?	2
40.	Write the four differences between physisorption and chemisorption?	2
41.	'Adsorption is an exothermic process '. Explain.	2
42.	What is the sign of ΔH , ΔS and ΔG when a gas is adsorbed by an adsorbent?	2
43.	(a) Which gas is adsorbed more readily on charcoal lump:	2
	ammonia or carbondioxide and why?	
	(b) Which adsorbs more of carbon monoxide:	
	charcoal lump or charcoal powder& why?	
44.	Name the factors which influence the extent of adsorption of a gas on solid.	2
45.	Explain Freundlich adsorption isotherm	2
46.	Define the terms – catalysis and catalyst.	2
47.	What is meant by activity and selectivity of a catalyst?	2
48.	What is shape – selective catalysis? Give an example of shape selective catalyst.	2
49.	Give two examples of enzyme catalysed reaction	2
50.	Explain the mechanism of enzyme catalysis.	2
51.	Differentiate between multimolecular and macromolecular colloid?	2
52	Write the equation for formation of	2
	(a) Sulphur sol.	
	(b) Ferric hydroxide sol.	
53.	How is a colloidal solution purified by dialysis?	2

hydrated ferric oxide sol:

54.	Define the terms – (a) Brownian movement (b) Electrophoresis? [2
55.	How does an emulsifying agent work? Give an example of emulsifying agent.	2
56.	Explain the effect of temperature on the extent of physical and chemicaladsorption.	2
57.	Define the term peptization and mention its cause.	2
58.	In what way these are different: (a) a sol and a gel (b) a gel and an emulsion.	2
59.	Define the terms:	2
	(a) Helmholtz electrical double layer.	
	(b) Zeta potential.	
60.	Mention the two necessary conditions for the observation of Tyndall Effect	2
61.	Account for the following:	2
	(a) Artificial rain can be caused by spraying electrified sand on the clouds.	
	(b) Electrical precipitation of smoke.	
62.	Write chemical equations for the preparation of sols:	2
	(a) Gold sol by reduction.	
	(b) Hydrated ferric oxide sol by hydrolysis.	